

8.22' The actual $\Delta T_{\text{realized}}$ is estimated to be about 0.6°C , which is 75% of the equilibrium ΔT

$$\Delta T_{\text{realized}} = 0.6^\circ\text{C} = 0.75 \Delta T_{\text{equilibrium}}$$

so, $\Delta T_{\text{equilibrium}} = 0.6/0.75 = 0.8^\circ\text{C}$

but, $\Delta T_{\text{equilibrium}} = \lambda \Delta F_{\text{actual}} = 0.57 \times \Delta F_{\text{actual}} = 0.8$

that is, $\Delta F_{\text{actual}} = \frac{0.8}{0.57} = 1.40 \text{ W/m}^2$

The direct forcing is 2.45 W/m^2 , so aerosols etc are $2.45 - 1.40 = 1.05 \text{ W/m}^2$

8.23 Energy sources and carbon intensity:

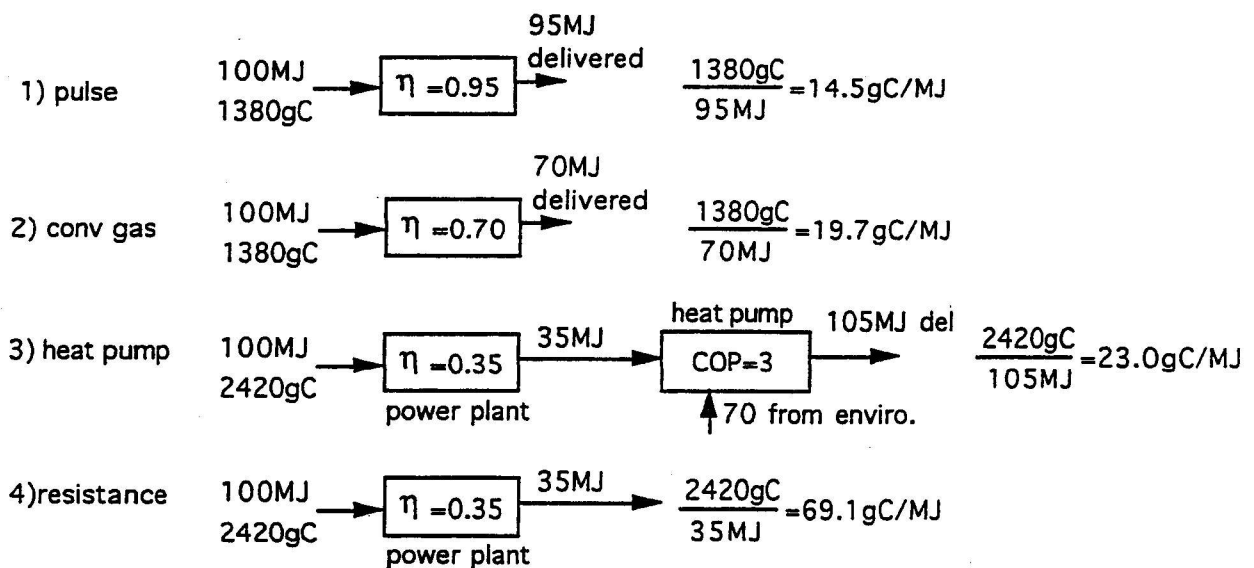
Coal	25%	@ 24.2 gC/MJ
Oil	45%	@ 19.7 gC/MJ
Gas	20%	@ 13.8 gC/MJ
Other	10%	@ 0

a. avg C intensity = $0.25 \times 24.2 + 0.45 \times 19.7 + 0.20 \times 13.8 + 0.10 \times 0 = 17.68 \text{ gC/MJ}$

b. Coal replaced by non-carbon emitting sources:

avg C intensity = $0.25 \times 0 + 0.45 \times 19.7 + 0.20 \times 13.8 + 0.10 \times 0 = 11.63 \text{ gC/MJ}$

8.27 Using LHV carbon intensities from Table 8.9, the four options are:



Notice the tremendous range: 14.5 to 69.1 gC/MJ, almost 5:1 !

8.34 Gasoline C_7H_{15} , 6.15 lbs/gal, fully combusted,

$$a. \text{ gasoline} = \frac{6.15 \text{ lb gas}}{\text{gal}} \times \frac{(7 \times 12 = 84) \text{ lbs C}}{(7 \times 12 + 15 \times 1 = 99) \text{ lb gas}} = 5.22 \text{ lbsC/gal}$$

$$C = \frac{40,000 \text{ miles}}{12 \text{ miles/gal}} \times \frac{5.22 \text{ lbs C}}{\text{gal}} = 17,394 \text{ lbsC that will be released}$$

b. 4000 lb car, 10,000 mi/yr,

$$C = \frac{17,394 \text{ lbsC}}{40,000 \text{ mi}} \times \frac{10,000 \text{ mi}}{\text{yr}} = 4348 \text{ lbs C/yr}$$

$$\frac{\text{carbon}}{\text{vehicle wt.}} = \frac{4348 \text{ lbs C/yr}}{4000 \text{ lb}} = 1.09$$

the car emits slightly more carbon per year than it weighs!

$$c. \text{ carbon tax} = \frac{5.22 \text{ lbs C}}{\text{gal}} \times \frac{\$15}{2000 \text{ lbs C}} = \$0.039/\text{gal} = 3.9\text{¢}/\text{gal}$$

d. new car @40mpg. for 40,000 miles:

$$C_{\text{reduction}} = 17,394 \text{ lbsC} - \frac{40,000 \text{ mi}}{40 \text{ mi/gal}} \times \frac{5.22 \text{ lbsC}}{\text{gal}} = 12,174 \text{ lbs C saved}$$

e. trading in the clunker for the 40 mpg vehicle would save

$$\text{tax savings} = 12,174 \text{ lbsC} \times \frac{\$15}{2000 \text{ lbsC}} = \$91 \text{ per car}$$

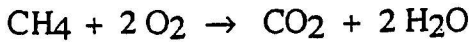
that is, those C offsets would save the utility \$91, which they could spend to get the clunker off the road.

8.33 Leaky landfill, 10 tonnes CH₄ per year:

a. 20-yr GWP for methane = 56,

$$10 \text{ tonnes CH}_4/\text{yr} \times 56 = 560 \text{ tonnes CO}_2(\text{equivalent})$$

b. burning the methane,



$$\begin{aligned} \text{CO}_2 \text{ emitted} &= \frac{1 \text{ mol CO}_2}{\text{mol CH}_4} \times \frac{(12 + 2 \times 16) \text{ tonne CO}_2 / \text{mol}}{(12 + 4 \times 1) \text{ tonne CH}_4 / \text{mol}} \times \frac{10 \text{ tonne CH}_4}{\text{yr}} \\ &= 27.5 \text{ tonne CO}_2 / \text{yr} \end{aligned}$$

c. equivalent CO₂ savings = 560 - 27.5 = 532.5 tonne CO₂

$$\text{as C: } 532.5 \text{ tonne CO}_2 / \text{yr} \times \frac{12 \text{ tonne C}}{44 \text{ tonne CO}_2} = 145.2 \text{ tonne C / yr saved}$$

d. carbon tax saved = 145.2 tonne C/yr x \$20/tonne C = \$2900 / yr

e. same thing, 532.5 tonne CO₂ saved x \$5.45/tonne CO₂ = \$2900/yr

8.35 Electrics vs gasoline powered cars:

$$\text{a. gas car emissions} = \frac{5.22 \text{ lbs C/gal}}{40 \text{ miles/gal}} \times \frac{1000 \text{ g}}{2.2 \text{ lbs}} = 59.3 \text{ gC / mi}$$

b. with the very efficient natural-gas-fired power plant,

$$\text{electric emissions} = \frac{8000 \text{ kJ}}{\text{kWh}} \times \frac{13.8 \text{ gC}}{\text{MJ}} \times \frac{\text{MJ}}{10^3 \text{ kJ}} \times \frac{\text{kWh}}{5 \text{ mi}} = 22.1 \text{ gC / mi}$$

c. with the typical old coal plant,

$$\text{coal plant heat rate} = \frac{1 \text{ kW heat in}}{0.30 \text{ kW electric out}} \times \frac{1 \text{ kJ/s}}{\text{kW heat in}} \times \frac{3600 \text{ s}}{\text{hr}} = 12,000 \text{ kJ / kWh}$$

$$\text{electric emissions} = \frac{12,000 \text{ kJ}}{\text{kWh}} \times \frac{24 \text{ gC}}{\text{MJ}} \times \frac{\text{MJ}}{10^3 \text{ kJ}} \times \frac{\text{kWh}}{5 \text{ mi}} = 57.6 \text{ gC / mi}$$

So, carbon can be saved with electric cars when efficient natural-gas power plants are assumed, but for the typical old coal plant, very little if any savings accrue.