

Substituent Effects in Electrophilic Aromatic Substitution Reactions

| substituent | name |
|--|---------------------|
| <u>activating and <i>ortho</i>-, <i>para</i>-directing</u> | |
| NH ₂ , NHR, or NR ₂ | amino |
| OH | hydroxyl (phenolic) |
| OR | alkoxyl |
| NHCOCH ₃ | acetamido |
| R | alkyl |
| Ar | aryl |
| <u>benzene itself (the reference standard)</u> | |
| H | |
| <u>deactivating and <i>ortho</i>-, <i>para</i>-directing</u> | |
| F | fluoro |
| Cl | chloro |
| Br | bromo |
| I | iodo |
| <u>deactivating and <i>meta</i>-directing</u> | |
| CHO | aldehyde |
| COOR | ester |
| COOH | carboxylic acid |
| COR | ketone |
| SO ₃ H | sulfonic acid |
| C≡N | nitrile |
| NO ₂ | nitro |
| ⁺ NR ₃ | ammonium |

Notes:

1. An **O** (in contrast to an O) denotes a carbonyl oxygen.
2. Substituents that are ranked higher in the list are more activating. Conversely, substituents that are ranked lower in the list are more deactivating.