## Chemistry 334R

## Problem Set 1

1. The ${ }^{1} \mathrm{H}$ and broadband proton-decoupled ${ }^{13} \mathrm{C}$ NMR spectra of compound $\mathrm{A}\left(\mathrm{C}_{7} \mathrm{H}_{12} \mathrm{O}_{2}\right)$ are shown below. The labels next to each of the resonances in the ${ }^{1} \mathrm{H}$ NMR spectrum signify the integrals and multiplicities observed ( $\mathrm{d}=$ doublet, $\mathrm{dd}=$ doublet of doublets, $\mathrm{dt}=$ doublet of triplets, sept $=$ septet. ) DEPT 90 and DEPT 135 data are included in the table on the following page. Use this spectroscopic evidence to determine the identity of the compound. Make clear assignments of all resonances to explain your reasoning.
${ }^{1} \mathrm{H}:$

broadband proton-decoupled ${ }^{13} \mathrm{C}$ :
193.1
146.8132 .5
71.170 .6
22.4

2. (continued)

${ }^{1} \mathrm{H}$ NMR assignments:
chemical shift (ppm)
9.68
6.84
6.26
4.04
3.19
1.16
${ }^{13} \mathrm{C}$ NMR assignments:
chemical shift (ppm)
193.1
146.8
132.5
71.1
70.6
22.4
assignment
explanation of multiplicity
3. Draw the major organic product that is formed from the following reaction. The broadband proton-decoupled ${ }^{13} \mathrm{C}$ NMR spectrum of the product is shown below. The labels next to each of the resonances signify the multiplicities observed in the corresponding off-resonance protondecoupled ${ }^{13} \mathrm{C}$ NMR spectrum ( $\mathrm{d}=$ doublet, $\mathrm{t}=$ triplet, $\mathrm{q}=$ quartet $)$. Use this spectroscopic evidence to determine the identity of the compound. Make clear assignments of all resonances to explain your reasoning.

Draw all the possible alternative products that are mechanistically reasonable. Describe how the spectroscopic data rules out each of these compounds as the actual product.

26.8
$135.6 \quad 124.3$
$31.7 \quad 24.0 \quad 14.3$

${ }^{13} \mathrm{C}$ NMR assignments:

## chemical shift (ppm) assignment explanation of multiplicity

135.6

## 124.3

31.7
26.8
24.0
14.3
3. Label all the sets of chemically equivalent protons for both of the following two compounds. Then draw annotated tree diagrams for protons $\mathrm{H}_{\mathrm{a}}$ and $\mathrm{H}_{\mathrm{f}}$. Specify, but do not quantify all the appropriate coupling constants in each tree diagram.
A.
B.



