From Chapter 17, The Sourcebook For Teaching Science Norman Herr, 2008

Estimate all of the values by expressing the following in Scientific Notation

(1)
$$\frac{1 \tan 100}{1 \tan 100} \frac{5124 \text{ sq} \text{ inches}}{1 \text{ syb6}} \frac{6.452 \text{ square cm}}{1 \text{ sq} \text{ inch}} \frac{0.0001 \text{ m}^2}{1 \text{ square cm}} = 3.3 \text{ m}^2$$

(2) $\frac{3.0 \times 10^8 \text{ m}}{\sqrt{s}} \frac{60 \text{ s}}{1 \text{ mn}} \frac{60 \text{ s}}{1 \text{ k}} \frac{60 \text{ mn}}{1 \text{ dgy}} \frac{24 \text{ k}}{1 \text{ sg}} \frac{365.25 \text{ dgy}}{1 \text{ sg}} = 9.47 \times 10^{15} \text{ m}$
(3) $\frac{1.0 \text{ g}}{2} \frac{0.0010 \text{ kg}}{1 \text{ mn}} \frac{0.001 \text{ tgnfie}}{1 \text{ kg}} \frac{0.9842 \text{ long tons}}{1 \text{ tgnfie}} \frac{12240 \text{ pounds}}{1 \text{ lgn ton}} \frac{16 \text{ ounces}}{138.3 \text{ carats}} = 4.9 \text{ carats}} = 4.9 \text{ carats}} = 4.9 \text{ carats}} \frac{10.2 \text{ g}}{1 \text{ Lg}^2} \frac{11.22 \text{ MB}}{1 \text{ lgg}} \frac{1024 \text{ MB}}{1 \text{ lgn}} \frac{1024 \text{ kB}}{1 \text{ lgn}} \frac{100 \text{ s}}{1 \text{ lgn}} \frac{1022 \text{ lgn}}{1 \text{ lgn}} \frac{100 \text{ s}}{1 \text{ lgn}} \frac{100 \text{ s}}{1 \text{ lgn}} \frac{100 \text{ s}}{1 \text{ lgn}} \frac{10 \text{ lgn}}{1 \text{ lgn}} \frac{10 \text{ lgn}}{1 \text{ lgn}} \frac{10 \text{ lgn}}{1 \text$

(2) $\frac{4.13 \text{ g}}{1.5 \text{ mL}} = \frac{2.8 \text{ g}}{\text{cm}^3}$ it is probably granite

(3) $\frac{y \text{ in } 100 \text{ cm } 1.\text{m}}{1.\text{ in } 2.54 \text{ cm } \text{ m}} = 39 \text{ y}$

(4) 93,000,000 miles 5280 ft 12 in $2.54 \text{ cm} = 1.4 \text{ x} 10^{13} \text{ cm}$

mile
$$ft$$
 in = 1.4 x 10 cm

$$\begin{array}{rcl} (5) & \frac{12.4}{1.64 \ cm^3} \left| 1 \ cm^2 \right| \frac{12.6}{mL} & \frac{7.56}{mL} & galena \\ & galena \ will float in mercury \\ \hline \frac{13.600 \ kg'}{m^3} & \frac{m^2}{1.0 \times 10^6 \ cm^3} \left| \frac{1 \ cm^2}{1 \ lmL} \right| \frac{1000 \ g}{kg} = \frac{13.6 g}{mL} & mercury \\ \hline \frac{13.600 \ kg'}{m^3} & \frac{m^2}{1.0 \times 10^6 \ cm^3} \left| \frac{1 \ cm^2}{1 \ lmL} \right| \frac{1000 \ g}{kg} = \frac{13.6 g}{mL} & mercury \\ \hline (6) & \frac{50.6 L}{0.6 \ correte} \left| 1 \times 10^6 \frac{V}{V} \right| \frac{1000 \ cm^3}{1 \ x 10^6 \ cm^3} = \frac{50.000,000 \ m^3 \ concrete} \\ \hline (7) & \frac{5.97 \times 10^{.24} \ kg'}{1 \ (6378 \ km^3)} \left| \frac{km'}{1000 \ ml} \left| \frac{km'}{1 \ x 10^6 \ cm^3} \right| \frac{m^2}{1 \ cm^3} \left| \frac{1 \ cm^3}{1 \ lmL} \right| \frac{1 \ cm^3}{kg} \right| \frac{1000 \ g}{g} = \frac{5.5 \ g}{mL} \\ \hline (1) & \frac{21.4 \ g}{1 \ mL} \left| 1 \ cm^4 \right| \frac{5.90 \ cm^3}{1 \ m^4} \right| \frac{12.00 \ cm^3}{1 \ cm^4} = 1.67272 \times 10^{.14} \ \mu g \\ \hline (3) & \frac{170 \ g}{100 \ cm^4 \ lmL} \left| \frac{5.90 \ cm^3}{V} \right| \frac{12}{1000 \ ml} \left| \frac{2.00 \ dm^4}{2.00 \ dm^4} \right| \frac{100 \ sale}{2.00 \ g} \ AgNO_3 \\ \hline (4) & \frac{58.4 \ g}{mole} \ AgCl} \left| \frac{2.00 \ mole}{NaCl} \left| \frac{V}{1 \ 1000 \ ml} \right| \frac{100 \ x 10^6 \ cm^3}{1 \ lmL} \right| \frac{100 \ x 10^6 \ cm^3}{L} \right| \frac{1.00 \ x 10^6 \ cm^3}{L} \\ \hline (4) & \frac{58.4 \ g}{100 \ g} \ Cl_2 \left| \frac{1.00 \ g}{2.00.0 \ g} \ MgCl_2 \right| \frac{1.50 \ x 10^6 \ cm^3}{1 \ lo00 \ ml} \ sole \ sole \ mole} \ MgCl_2 \ = 1.40 \ Mcl_2 \ \frac{1.50 \ x 10^6 \ cm^3}{R} \\ \hline (5) & \frac{106 \ MgCl_2 \ 20.0 \ g}{1.20 \ mole} \ AgRCl_2 \ \frac{1.50 \ x 10^6 \ cm^3}{R} \right| \frac{1.00 \ x 10^6 \ cm^3}{L} \\ \hline (7) & V = \frac{mass}{density} \ density = \frac{mass}{V} \ V_{ubre} = V_{olutior} = 100 \ g \ solute \\ \hline (7) & V = \frac{mass}{density} \ density = \frac{mass}{V} \ V_{ubre} = V_{olutior} = 1.40 \ mole \ MgCl_2 \ = 1.40 \ Mcl_2 \ \frac{1}{R} \ ml^2} \\ r = \frac{8.25 \ cm^3}{10 \ cm^3} \left| \frac{1 \ cm^3}{10 \ cm^3} \left| \frac{1.50 \ x \ 10^6 \ g}{1.20 \ g^5} \left| \frac{1.00 \ x \ 100 \ g}{1.20 \ g} \left| \frac{1.00 \ x \ 100 \ g}{1.20 \ g} \left| \frac{1.00 \ x \ 100 \ g}{1.20 \ g} \left| \frac{1.00 \ x \ 100 \ g}{1.20 \ g} \left| \frac{1.0 \ x \ 10^6 \ cm^3}{10 \ m^2} \left| \frac{1.0 \ x \ 10^6 \ m^3}{10 \ m^2} \left| \frac{1.0 \ x \ 10^6 \ m^3}{10 \ m^3} \left| \frac{1.0 \$$